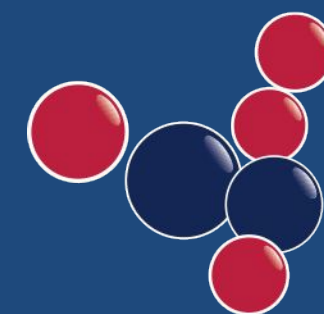


# Nitrate Removal from Potable Water

Kaitlyn Clark  
March 31, 2021



**RESINTECH<sup>®</sup> INC.**

INNOVATIONS IN ION EXCHANGE

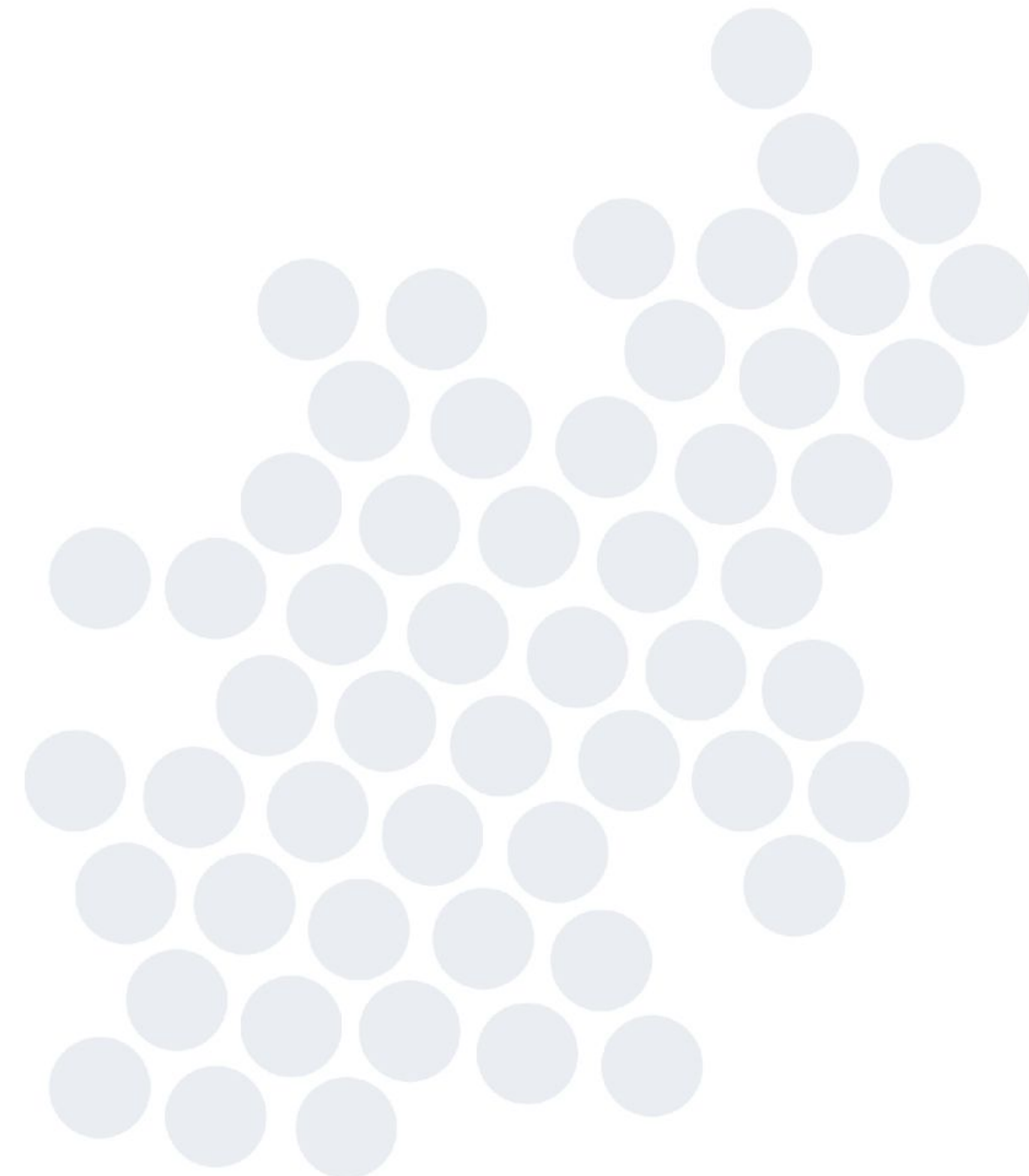
# How do nitrates get into our water?





# Outline

- History of Regulations
- Nitrate Chemistry
- Methods of Nitrate Removal
- Ion Exchange Theory
- Focus on Ion Exchange
- Closure
- Q & A



# How Nitrates Get into Water

- Surface runoff from agricultural areas
- Livestock
- Non-agricultural sources of nitrate
  - Lawn fertilizers
  - Septic systems
  - Domestic animals in residential areas

# Health Effects

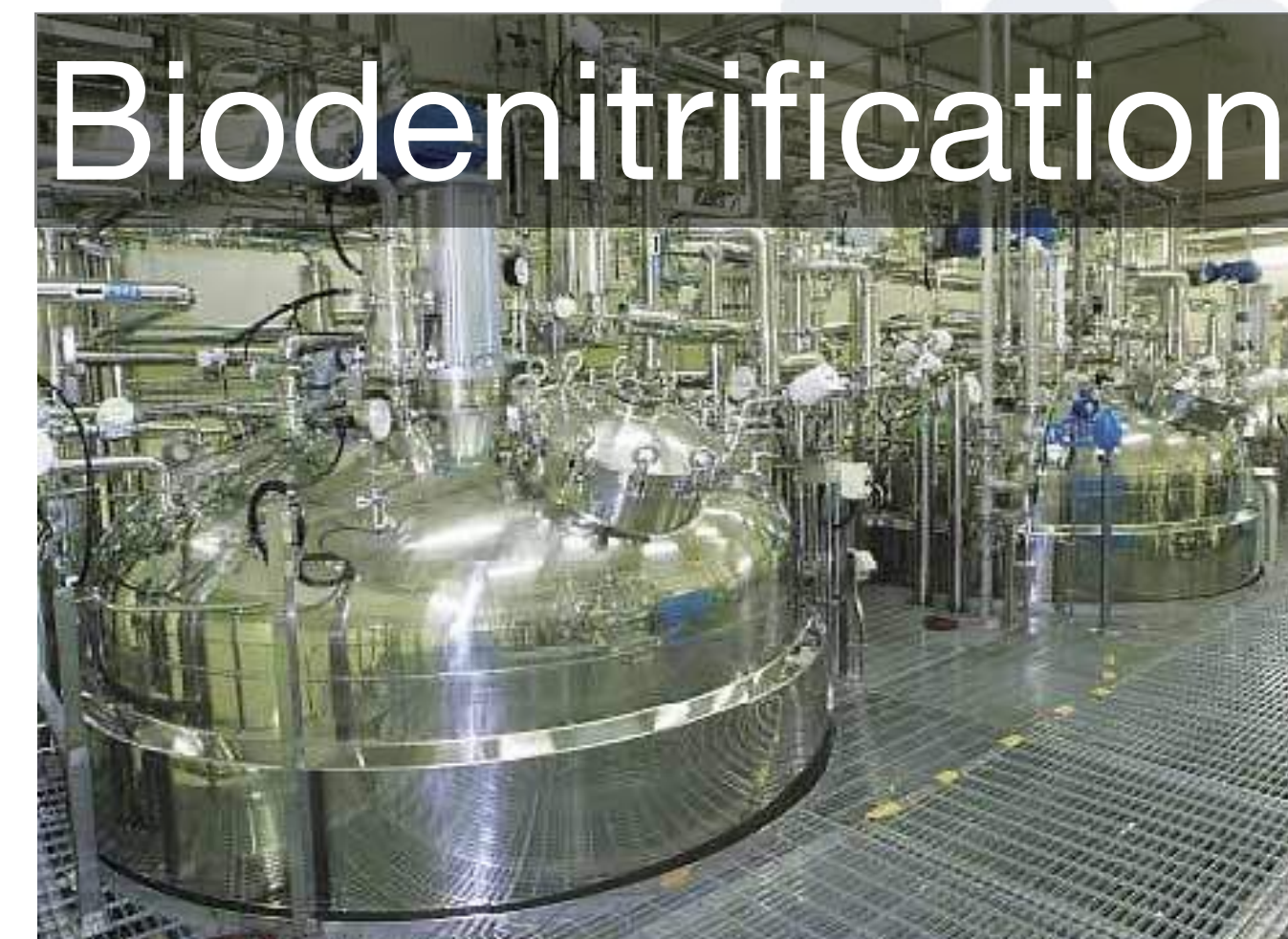
## Nitrate and Nitrite

- **Primarily**
  - Methemoglobinemia  
(Fetuses and infants <6 months old are most at risk)
- **Secondarily**
  - Nutrient Source for Harmful Algae Blooms (HAB)

# Nitrate Facts

- Nitrate salts are freely soluble
- Symmetrical ion is difficult to break apart
- Nitrate and uranium are sometimes co-contaminants
- IX resins that remove nitrate also remove uranium
- MCL is 10 mg/L as N (35.7 as  $\text{CaCO}_3$  and 44.7 as  $\text{NO}_3$ )

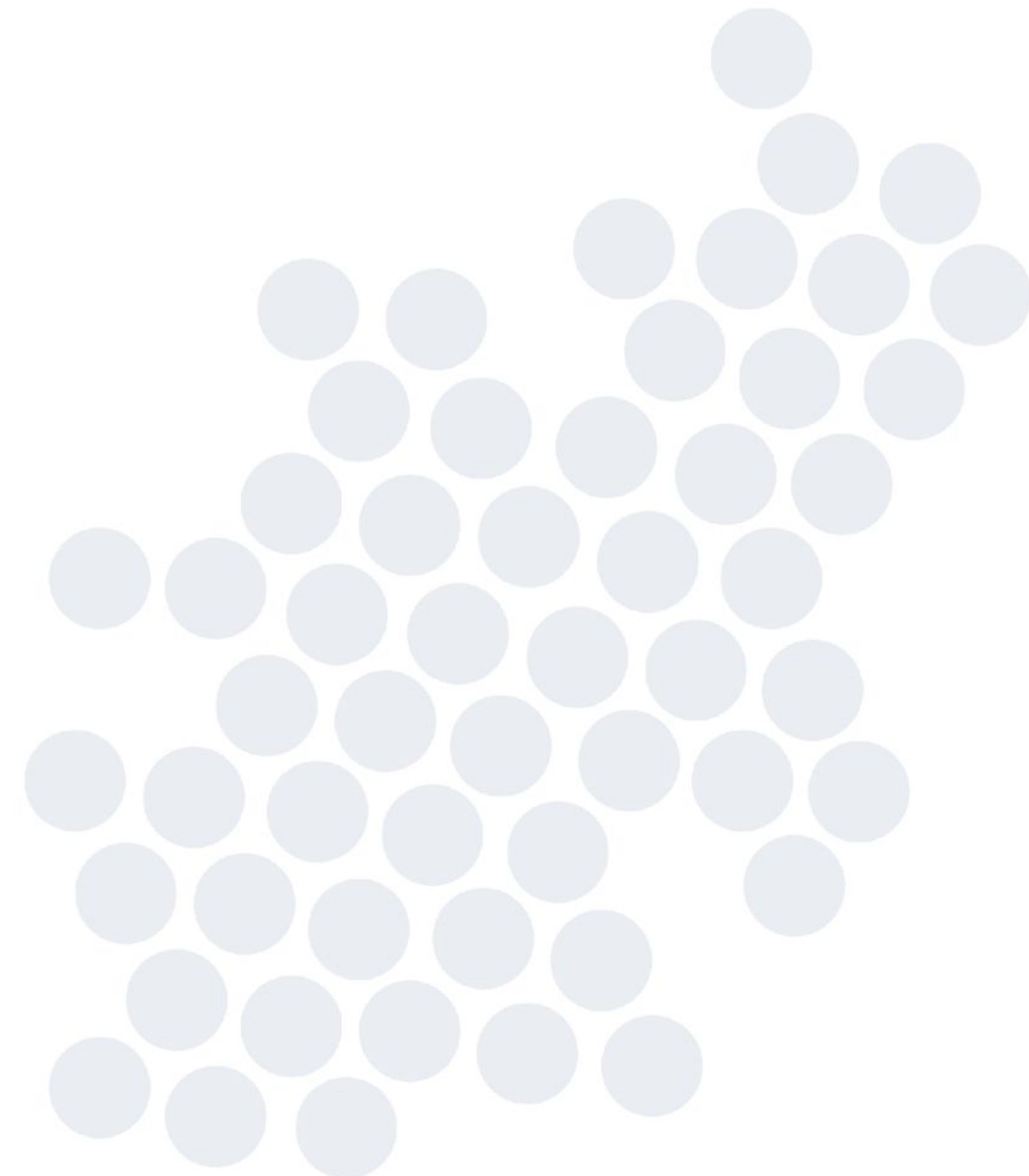
# Methods of Nitrate Removal



# Focus on Ion Exchange



- Simplified IX Theory
- Choosing the right resin
- Basic sizing
- Resin life
- Regeneration strategies
- Unexpected consequences





# Simplified Ion Exchange Theory

Ion exchange resins are **plastic beads** that **take salt out of water** and **put other salts back in.**

**ResinTech SIR-100-HP**

Nitrate Selective Anion Exchange Resin



# Nitrate Reduction by Anion Exchange



What else do we need to know?

- TDS or conductivity
- pH
- Hardness
- Sulfate
- Nitrate
- Chloride
- Alkalinity



# ResinTech Water Testing Kit



# Nitrate Removal Resins

| Category<br>Type          | DMEA<br><i>Type II</i> | TEA<br><i>Nitrate Selective</i> | TBA<br><i>Super Selective</i> |
|---------------------------|------------------------|---------------------------------|-------------------------------|
| ResinTech product         | SBG2-HP                | SIR-100-HP                      | SIR-110-HP                    |
| “K” (NO <sub>3</sub> /Cl) | 3                      | 5                               | 23                            |
| “K” (SO <sub>4</sub> /Cl) | 0.15                   | 0.02                            | 0.003                         |
| Capacity<br>(meq/mL)      | 1.45+                  | 0.95                            | 0.75                          |

DMEA - DIMETHYLETHANOLAMINE

TEA - TRIETHYLAMINE

TBA - TRIBUTYL AMINE

# Nitrate Removal Versus Softening

| Nitrate Removal                         | Hardness Removal                                      |
|---|---|
| K(NO <sub>3</sub> /Cl) remains constant | K(hardness/Na) is high at low TDS and low at high TDS |
| Needs high salt dose                    | Needs low salt dose                                   |
| High nitrate leakage in service cycle   | Low hardness leakage in service cycle                 |

# Choosing a Nitrate Removal Resin

| Type                            | DMEA     | TEA        | TBA        |
|---------------------------------|----------|------------|------------|
| ResinTech Equivalent            | SBG2-HP  | SIR-100-HP | SIR-110-HP |
| Cost                            | Low      | Higher     | Highest    |
| Nitrate Dumping                 | Possible | Rare       | None       |
| Capacity - Low SO <sub>4</sub>  | Highest  | Lower      | Lowest     |
| Capacity - High SO <sub>4</sub> | Low      | Higher     | Highest    |

# Basic Sizing Guidelines

Flow Rate is Key (based on average flow)

|  | Municipal | Residential / POE |
|--|-----------|-------------------|
| <b>Linear Flow Rate</b><br>(gpm / ft <sup>2</sup> )  | 10        | 10                |
| <b>Service Flow Rate</b><br>(gpm / ft <sup>3</sup> ) | 2 - 4     | 2 - 4             |
| <b>Min. Bed Depth</b><br>(Inches)                    | 60"       | 24"               |

# Residential

## Point of Entry System (POE)



## Point of Use System (POU)





# Municipal



# Municipal



# Design Guidelines

## Resin Requirements

| Parameter              | Value         | Reason   |
|------------------------|---------------|--|
| Min. Run Time          | 4 hours       | Shorter Cycle = Shorter Lifetime               |
| Min. Service Flow Rate | 0.5 gpm/cu.ft | Too Slow = Risk of Channeling                  |
| Max. Service Flow Rate | 5 gpm/cu.ft   | Too Fast= Inadequate Contact Time for Exchange |
| Min. Linear Flow Rate  | 2 gpm/sq.ft   | Too Slow = Risk of Channeling                  |
| Max. Linear Flow Rate  | 20 gpm/sq.ft  | Too Fast= Excessive Pressure Drop              |

# Resin Cycle Time and Lifetime Cycles

Average resin life is ~3,000 cycles

1 cycle / day



8 years

4 cycles / day



2 years



# Loss of Capacity over Time

- All anion resins lose capacity over time
- Physical losses ~1% per year
- Chemical losses - 2 to 5% / yr. for chloride form anion resins
- Fouling (increases apparent capacity loss)
  - Organics
  - Iron, manganese

# Factors affecting Resin life

- Chlorine in feedwater
- Frequency of regeneration
- Feedwater has foulants
- Feedwater is biologically active

# ResinTech Lab Services



# Examples of Nitrate Removal Projections

**In all cases we fixed the following:**

Chloride = 100 mg/L as Cl

Alkalinity = 50 mg/L as  $\text{HCO}_3$

Nitrate = 20 mg/L as N

**Sulfate was varied to illustrate differences in resin performance**

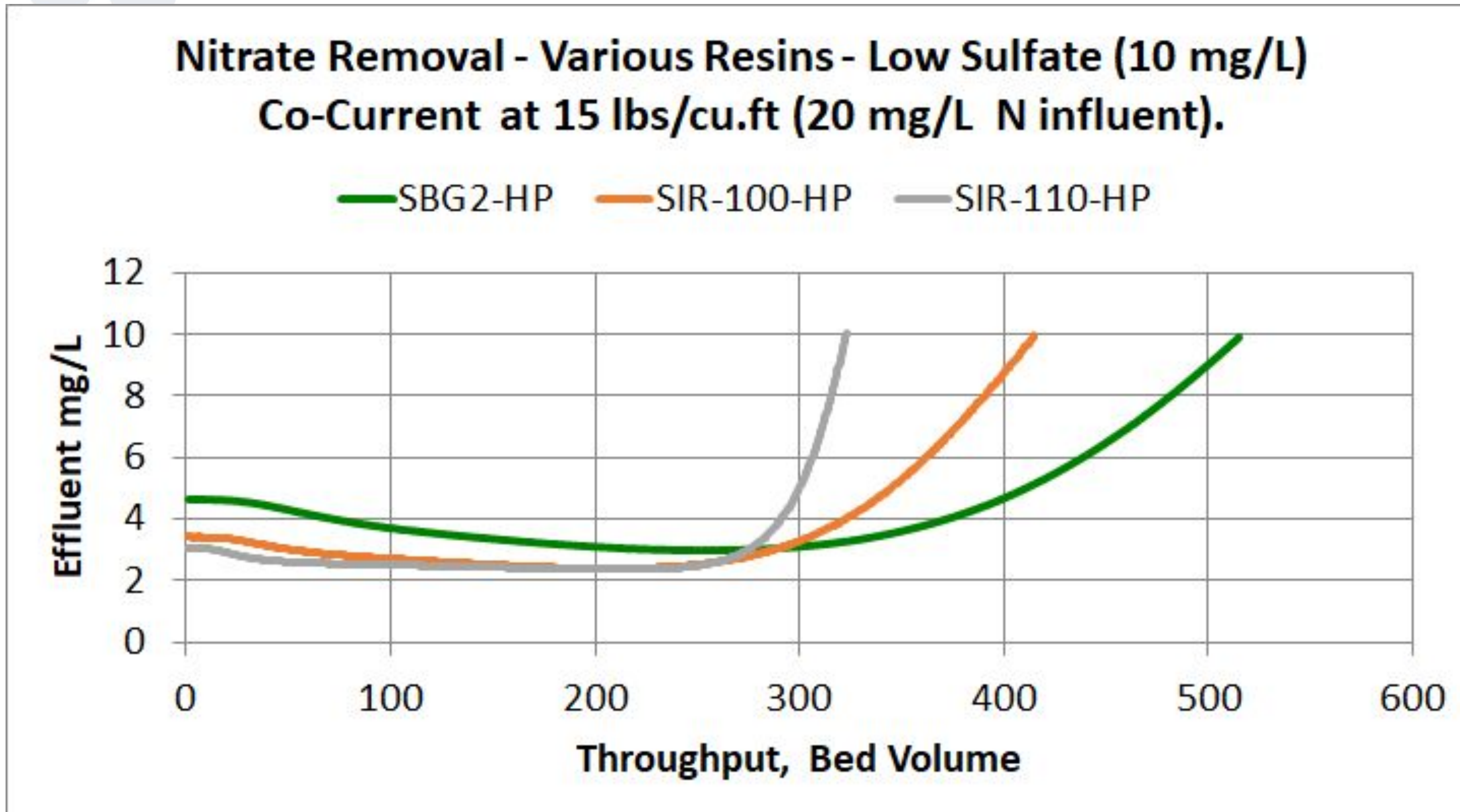
Low sulfate = 10 mg/L as  $\text{SO}_4$

Medium sulfate = 100 mg/L as  $\text{SO}_4$

High sulfate = 500 mg/L as  $\text{SO}_4$

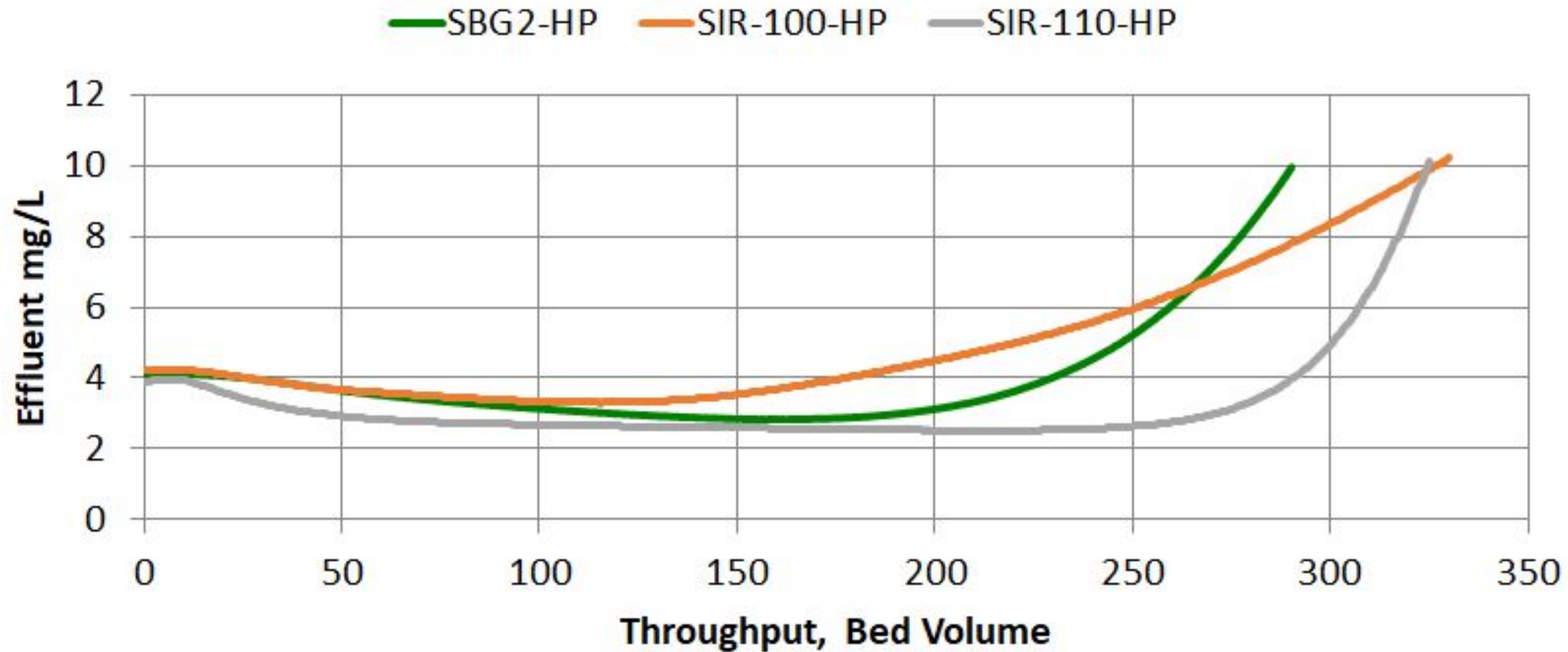


# Low Sulfate (10 mg/L as SO<sub>4</sub>)

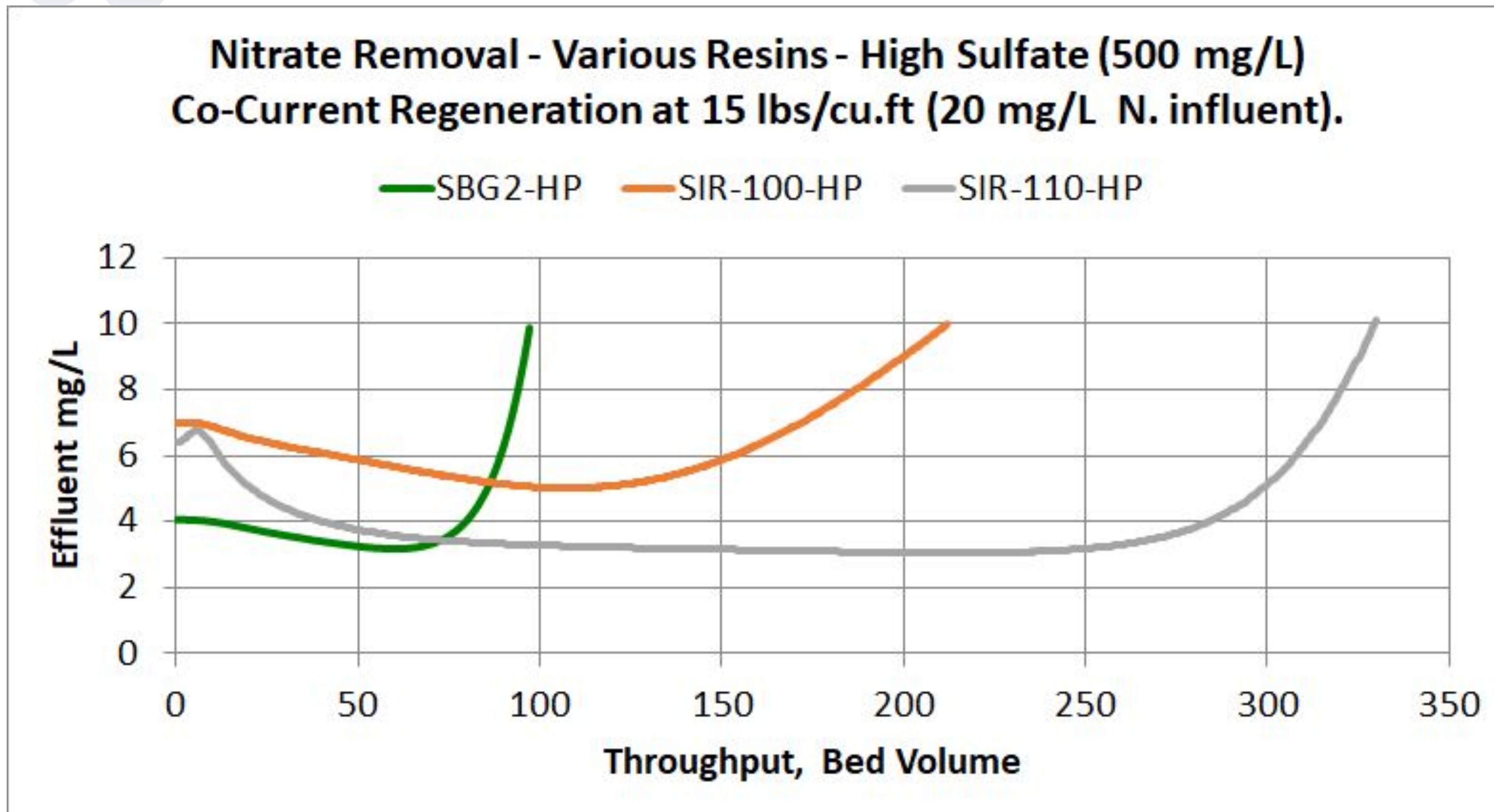


# Moderate Sulfate (100 mg/L SO<sub>4</sub>)

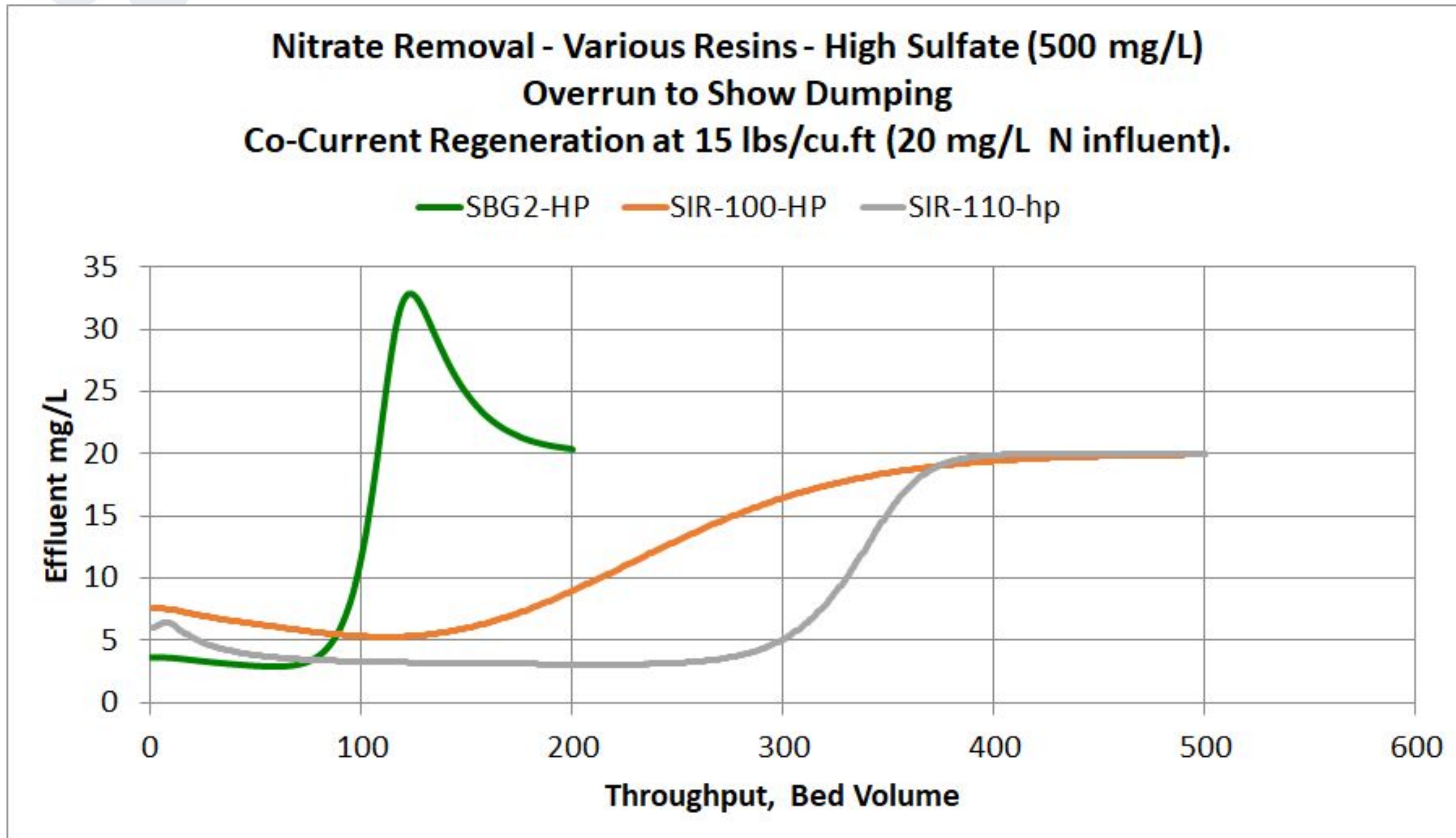
Nitrate Removal - Various Resins - Moderate Sulfate (100 mg/L)  
Co-Current Regeneration at 15 lbs/cu.ft. (20 mg/L N influent)



# High Sulfate (500 mg/L as SO<sub>4</sub>)



# Overrun High Sulfate (500 mg/L as SO<sub>4</sub>)



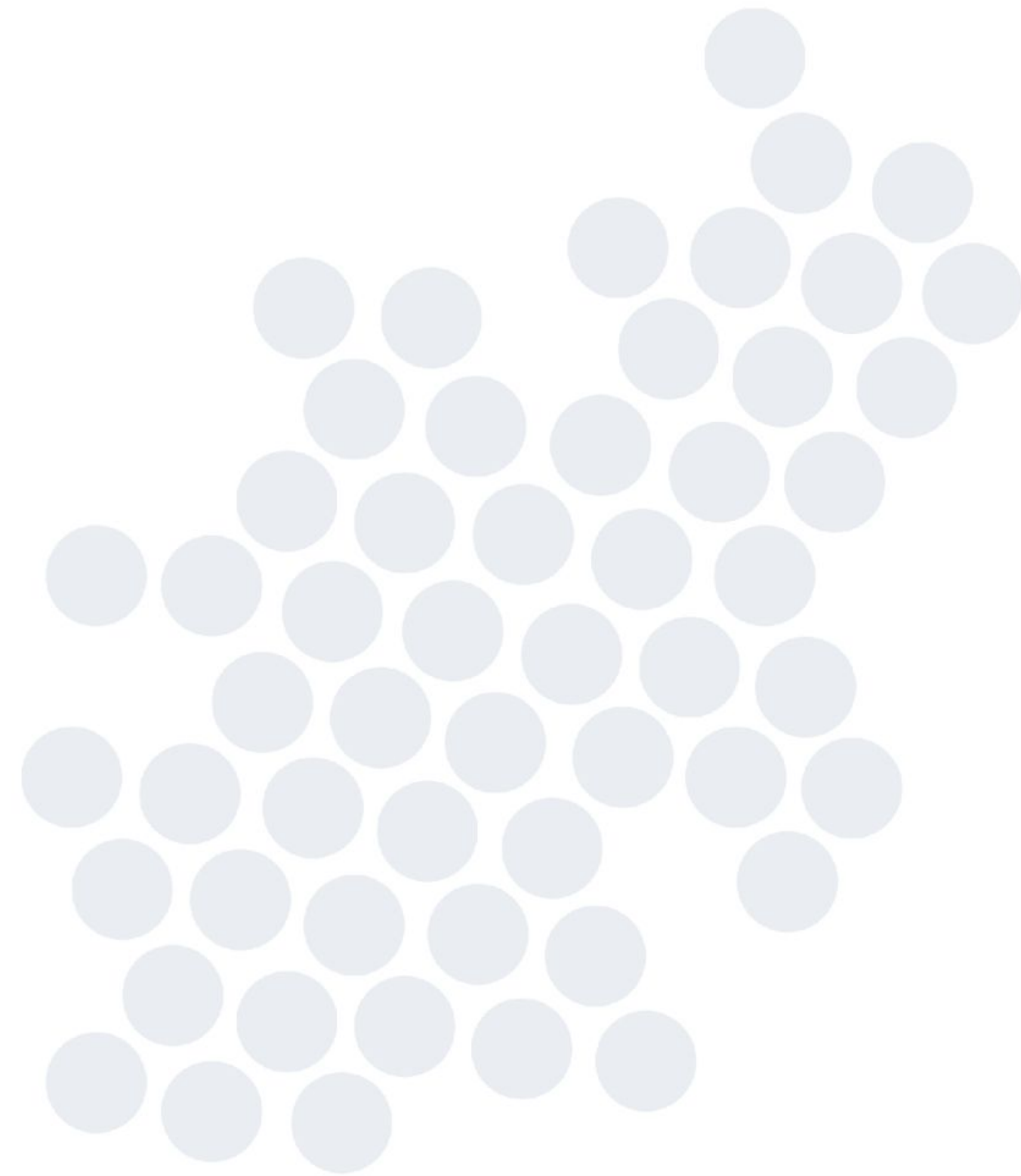
# Co-flow vs Countercurrent Regeneration

## Co-flow

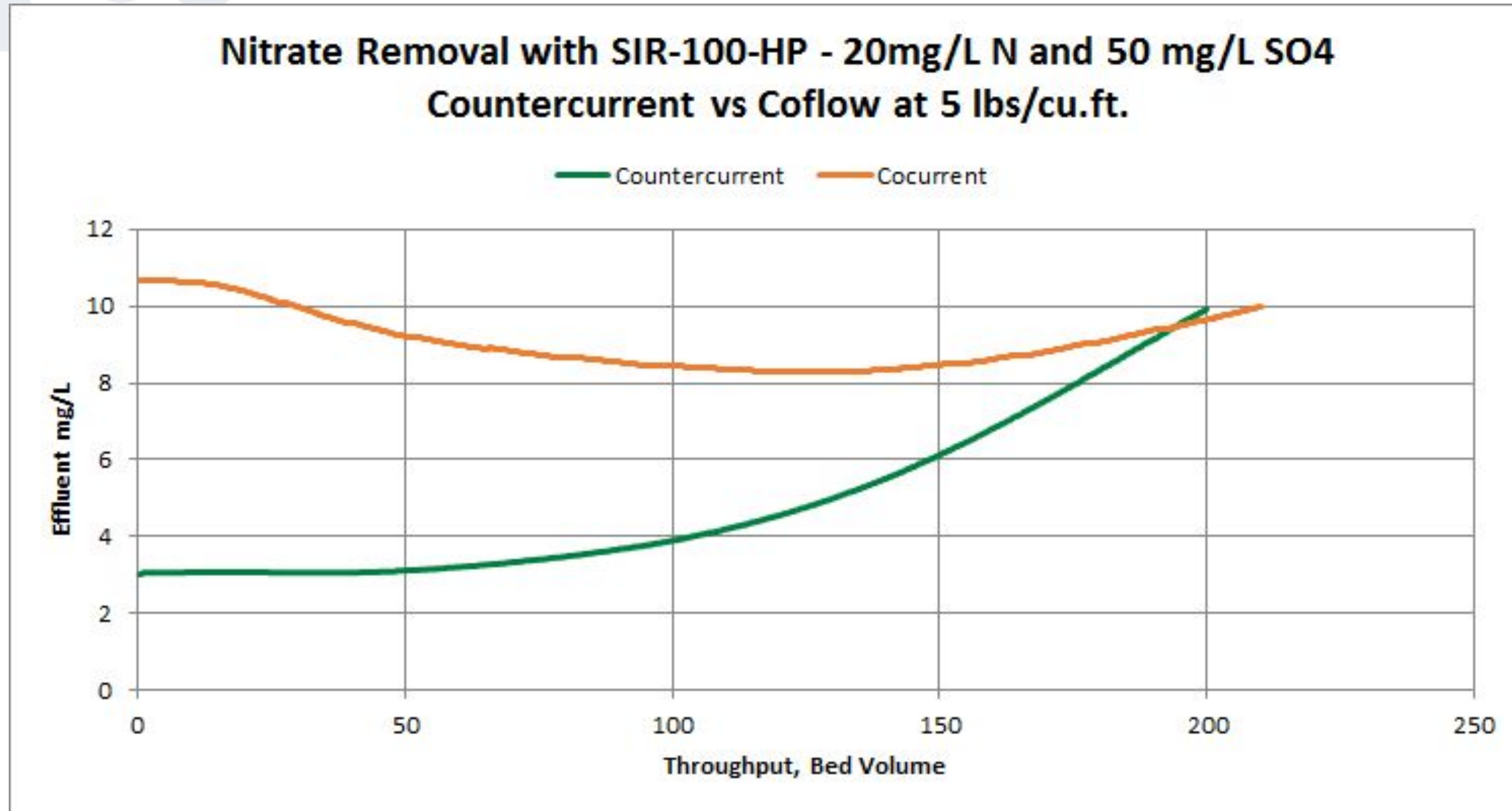
- Best known
- Most forgiving
- Least efficient

## Countercurrent

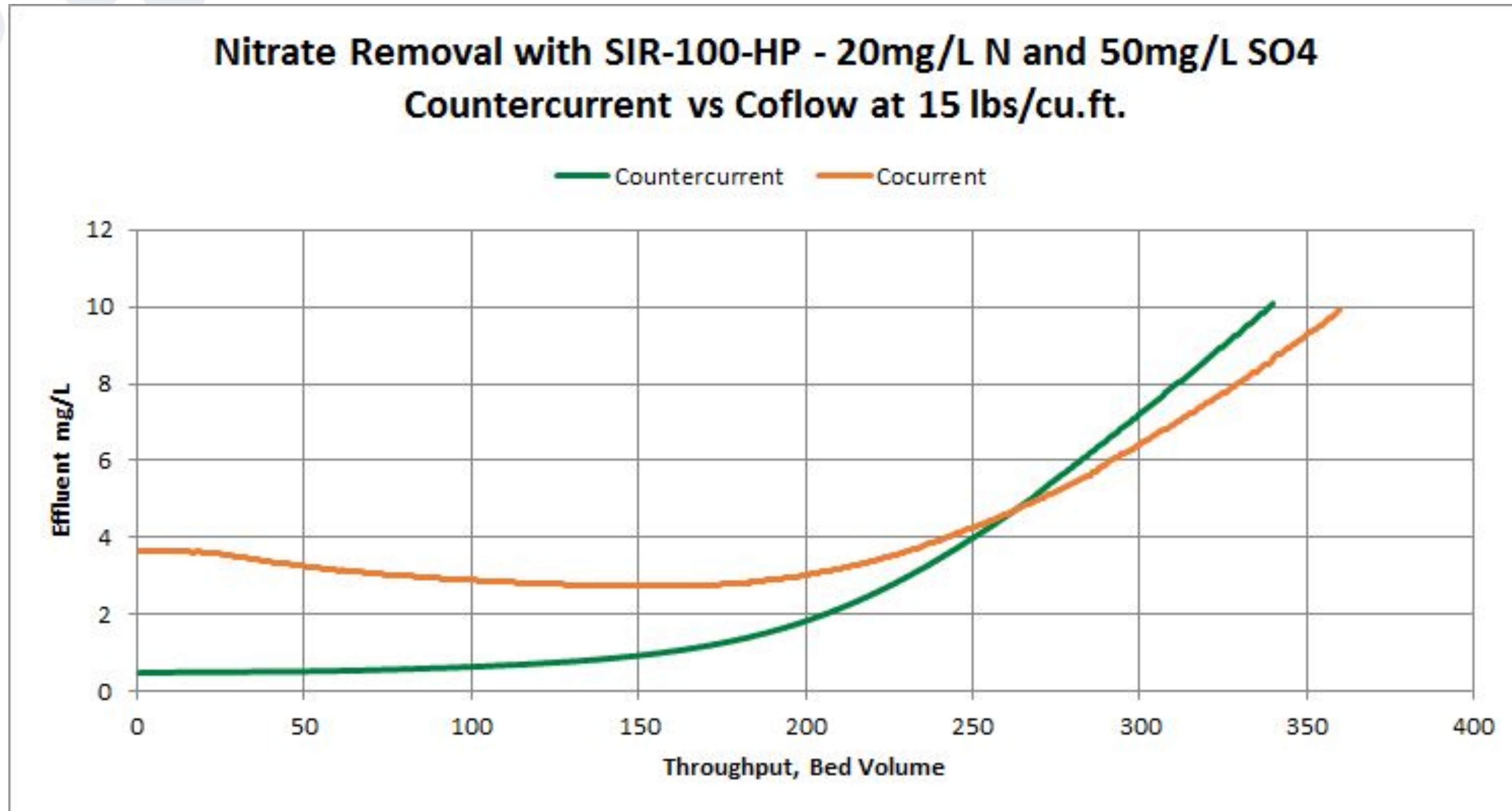
- More complicated (except packed beds)
- Higher salt efficiency (at low salt doses)
- Lower nitrate leakage



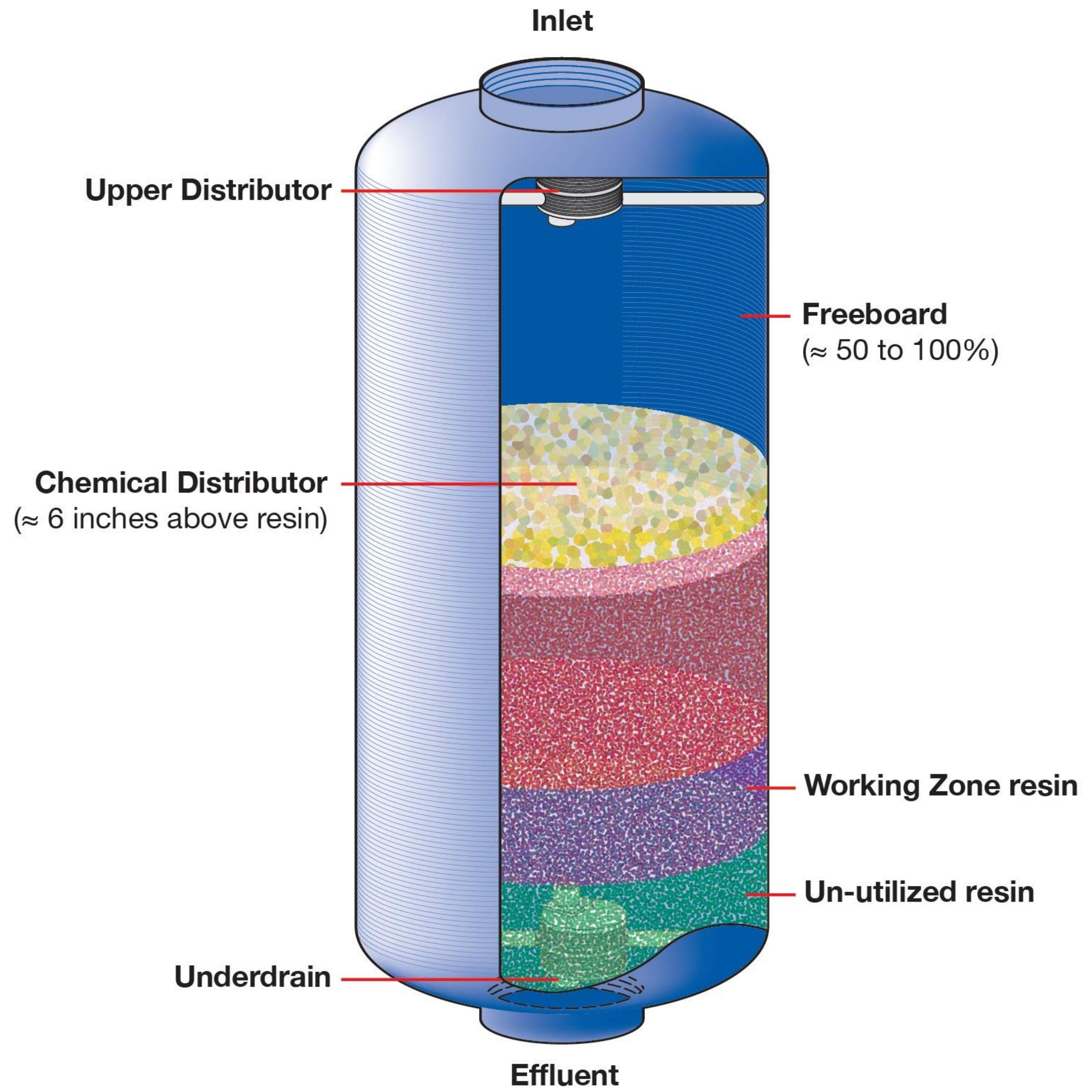
# CCR vs Coflow @ 5 lbs/cu.ft. Regeneration Dose



# CCR vs Coflow @ 15 lbs/cu.ft. Regeneration Dose



# Co-Flow Ion Exchanger



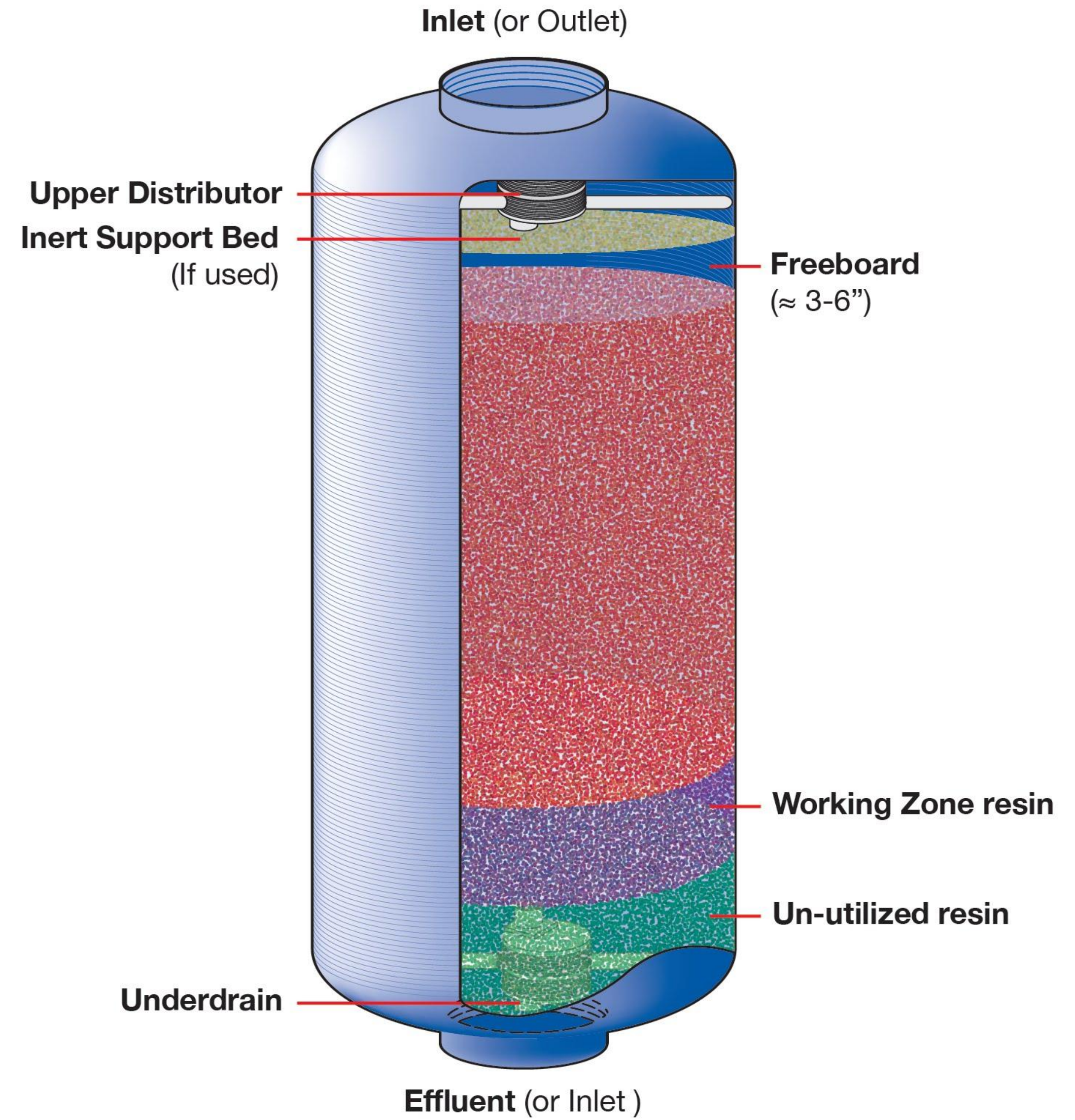


# Co-Flow Design

## Countercurrent Regeneration

| Pros            | Cons                             |
|-----------------|----------------------------------|
| Simplest        | relatively inefficient           |
| Least expensive | higher leakage in high TDS water |

# Packed Bed Ion Exchanger



# Packed Bed Design

## Countercurrent Regeneration

| Pros            | Cons                    |
|-----------------|-------------------------|
| high efficiency | more complex            |
| low leakage     | more expensive          |
|                 | needs external backwash |

# Developing a Regeneration Schedule

All Regeneration Schedules Must Include:

- Backwash (usually ~15 minutes)
- Chemical injection (usually ~30 mins)
- Slow/Displacement Rinse (Target 10-15 gallons/cu.ft)
- Fast Rinse (Target 30-40 gallons/cu.ft)
- Whole Kit and Kaboodle usually takes about 60-90 minutes

# Is Softening Necessary?

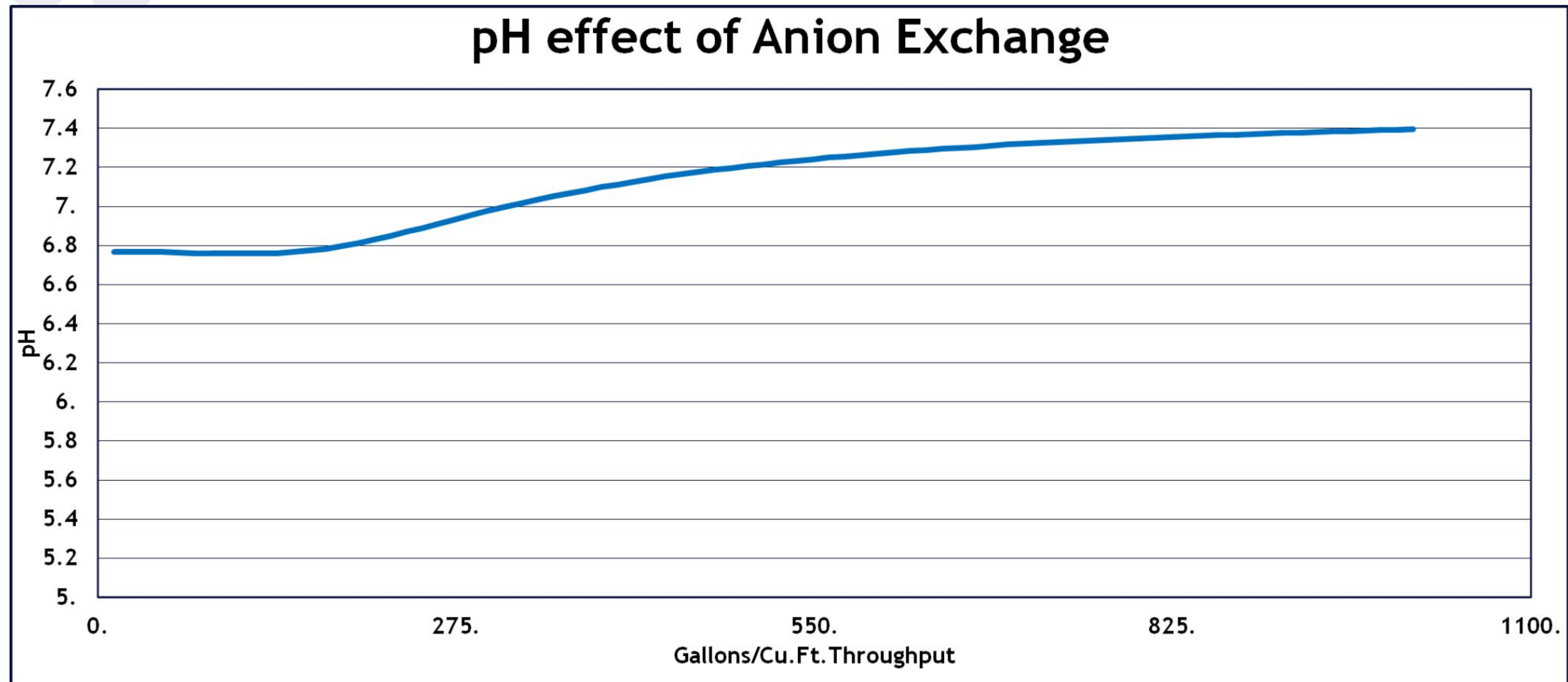
- Exhaustion cycle = sometimes when softening is needed for other reasons
- Regeneration Water = sometimes

# Ways to minimize scaling potential in the brine

- Buy soft salt (expensive and sometimes difficult to source)
- Soften dilution water
- Reduce brine concentration (or use two step brine injection)
- Increase brine flow rate
- Nitrate and super nitrate selective resins ([ResinTech SIR-100](#) and [SIR-110](#)) are less susceptible to brine scaling than type 1 or type 2 resins

# pH Change

from Chloride Form Anion Exchange Resin



# Types of Fouling

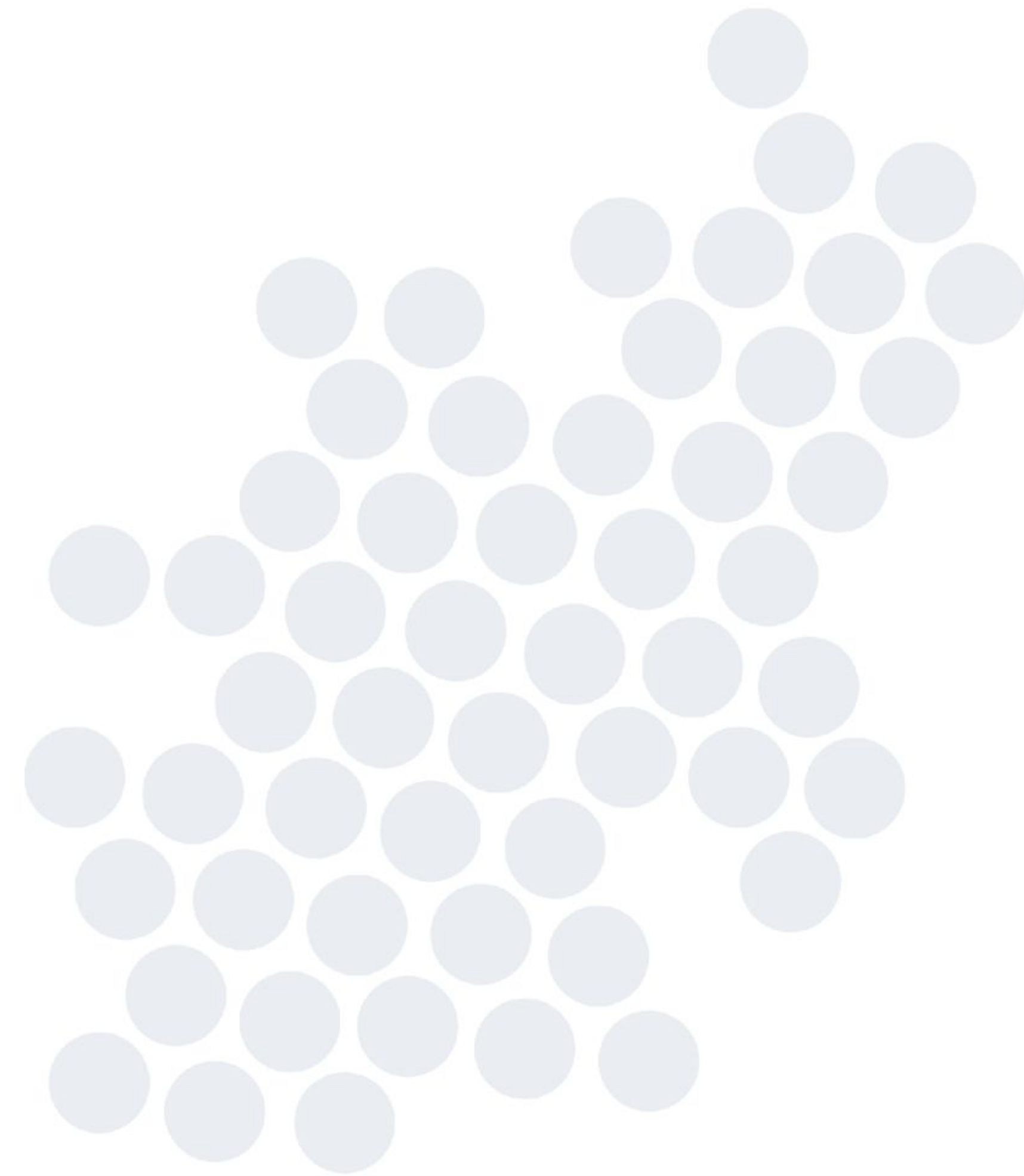
- **The plastic bag**
  - Anything that coats the beads
  - blocks the ions from getting in or out
- **The sponge**
  - Anything that fills up the spaces among the beads
  - Prevents equal flow
- **Super ions**
  - Any ion with very high selectivity
  - Blocks ion exchange sites

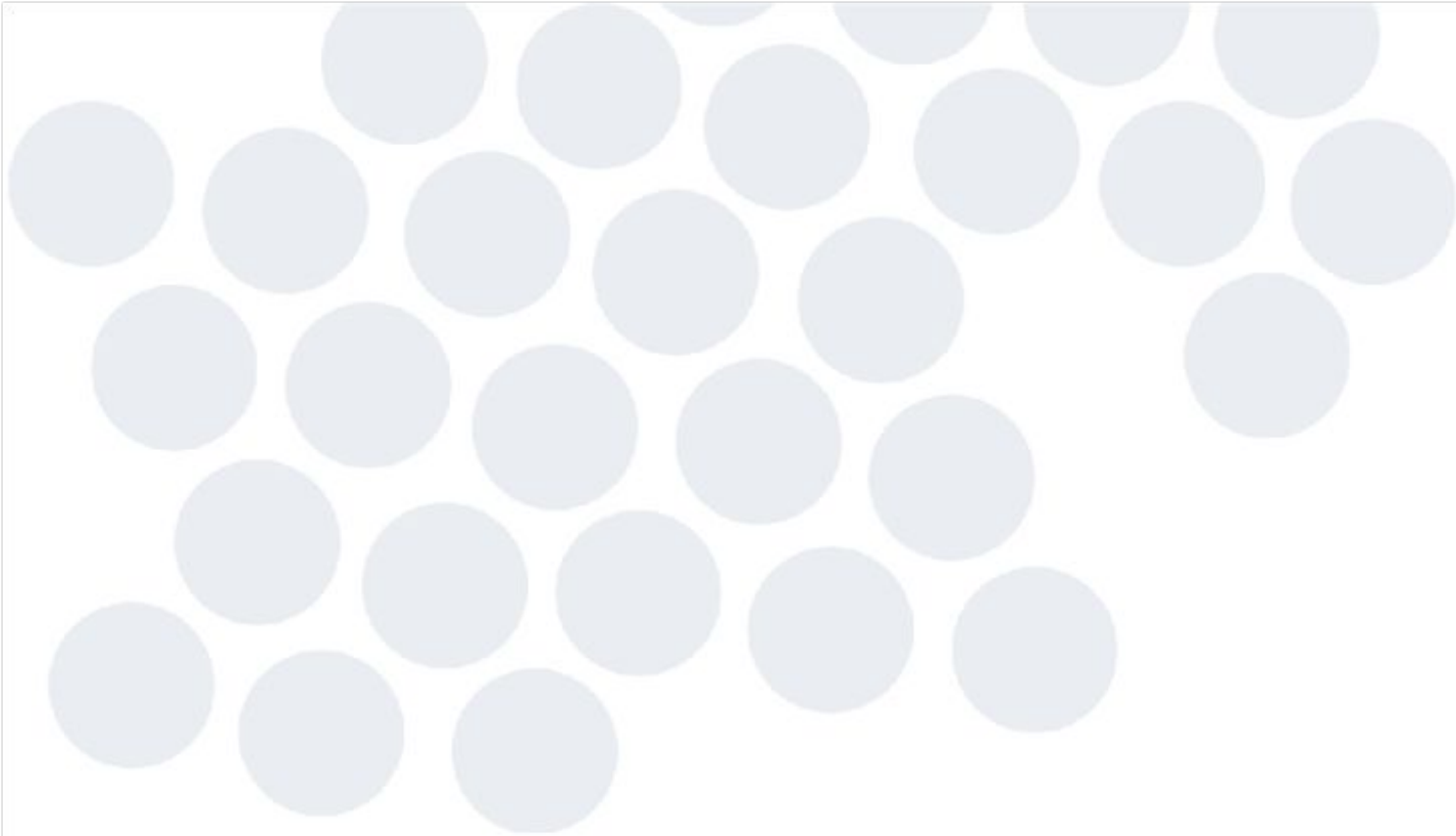




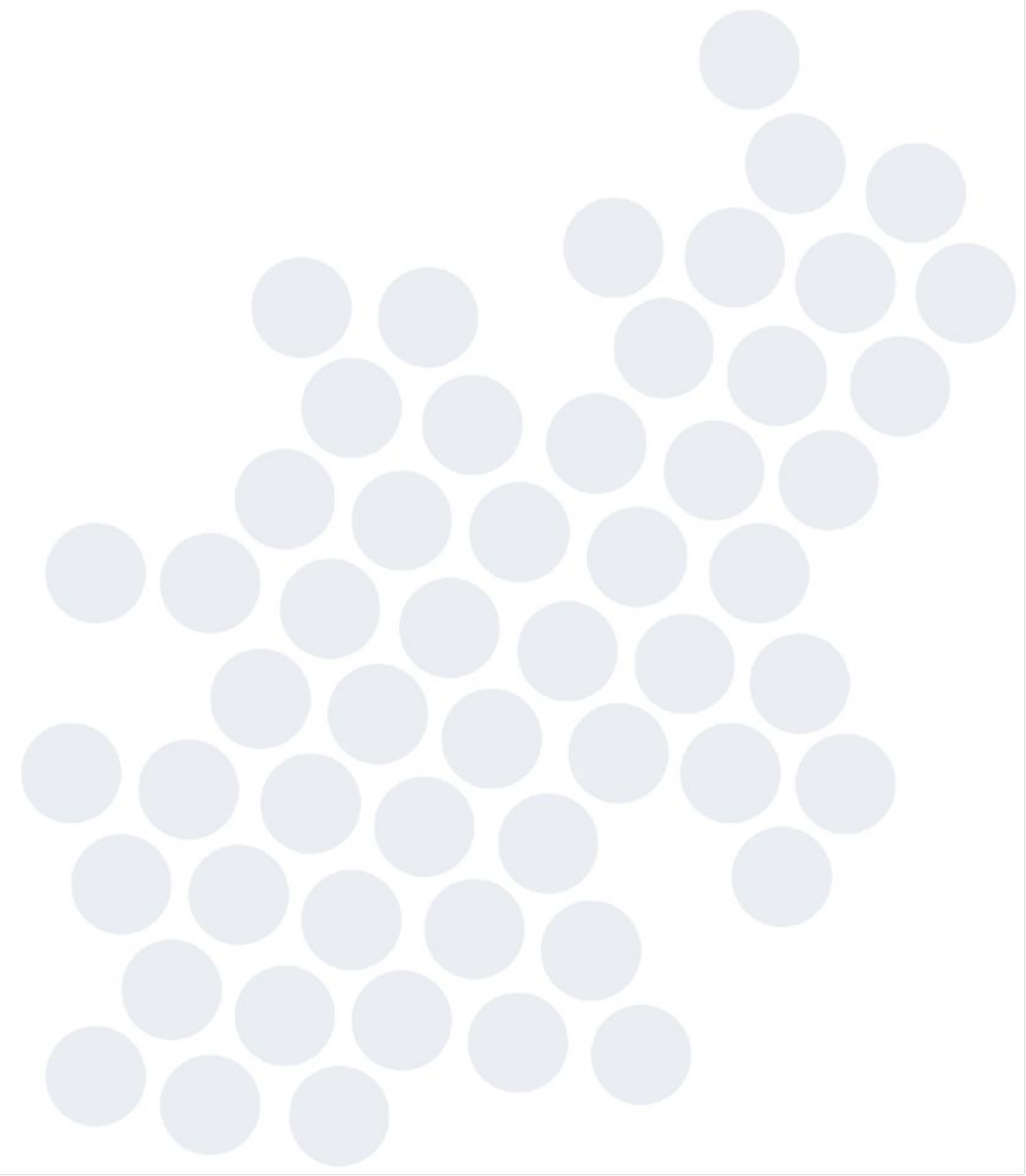
# Takeaways

- Water analysis is key
- Sulfate is king
- Regular “wellness checks”





Questions?

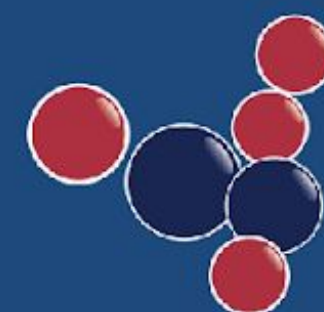


# THANK YOU

**Kaitlyn Clark**

p. **856-422-5224**

e. **[kclark@resintech.com](mailto:kclark@resintech.com)**



**RESINTECH<sup>®</sup> INC.**

INNOVATIONS IN ION EXCHANGE